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$$\sum_{i=1}^c z_i(1 - K_i) = Y \sum_{i=1}^c z_i(K_i - 1)$$

Where Y is the fraction of feed that vaporizes ($Y = \text{vapor/feed}$), z_i is the mole fraction of the i th component in the feed,

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and K_i is the phase equilibrium ratio, commonly simply referred to as the K -value ($K_i = y_i/x_i$).

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